Badji Mokhtar University of Annaba Faculty of Medicine

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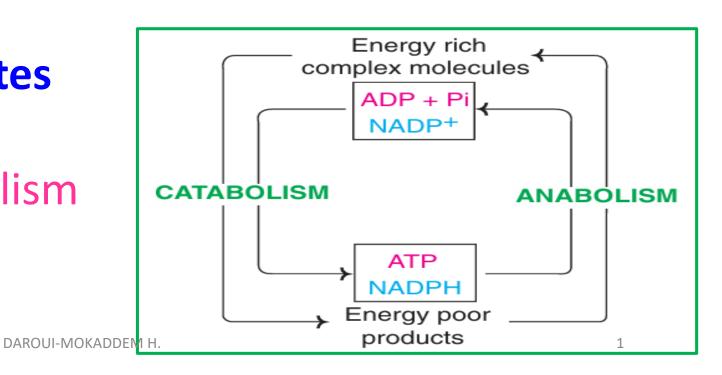
2025/2026

**MODULE: BIOCHEMISTRY** 

**CHAPTRE 1: Carbohydrates** 

**B-** Carbohydrate Metabolism

DAROUI-MOKADDEM H.



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#### Introduction

After digestion and absorption, nutrients undergo a series of biochemical transformations collectively referred to as intermediate metabolism. This term encompasses all intracellular enzymatic reactions responsible for the conversion, interconnection, and regulation of molecules to maintain the cell's energetic balance.

#### Metabolic pathways are classified into three main types:

- Anabolic pathways: These include synthetic reactions that build complex molecules (such as glycogen, lipids, and proteins) from simple precursors. These processes require energy input, provided by ATP or reduced coenzymes.
- ➤ Catabolic pathways: These involve oxidative degradation reactions of energy substrates (carbohydrates, lipids, amino acids), leading to the release of energy in the form of ATP, NADH,H<sup>+</sup>, or FADH<sub>2</sub>. These energy carriers subsequently feed the Electron respiratory chain (ETC).
- Amphibolic pathways: These are dual-purpose pathways that function as **intermediates** between **anabolism** and **catabolism**. The Krebs cycle (citric acid cycle) represents the principal metabolic crossroads of this type.

**Note:** Understanding **normal** carbohydrate **metabolism** is essential for interpreting **metabolic disorders** associated with various **pathologies**. Diabetes mellitus is a typical example, characterized by impaired regulation of blood glucose and energy metabolism.

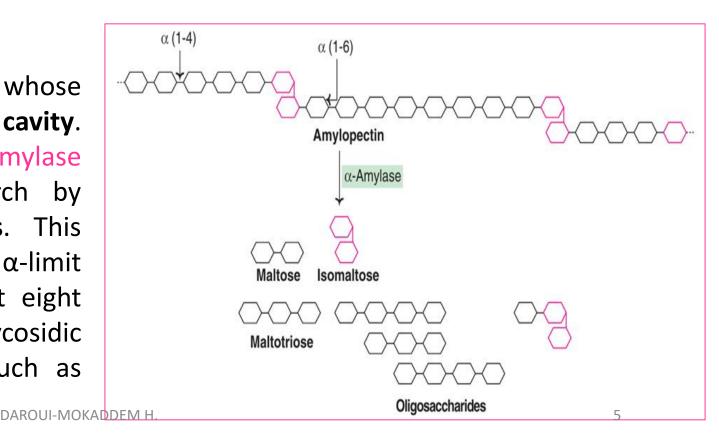
# Digestion and Absorption of Dietary Carbohydrates

### **Digestion**

- The principal dietary carbohydrates are **polysaccharides** (starch, glycogen), **disaccharides** (lactose, sucrose) and, to a minor extent, **mono-saccharides** (glucose, fructose).
- The digestion of carbohydrates begins in the mouth and is completed primarily in the small intestine.

#### **Digestion in the mouth:**

Carbohydrates are the only macronutrients whose digestion **begins** significantly in the **oral cavity**. During mastication, the enzyme salivary  $\alpha$ -amylase (ptyalin) initiates the hydrolysis of starch by randomly cleaving  $\alpha$ -1,4-glycosidic bonds. This enzymatic activity results in the formation of  $\alpha$ -limit **dextrins** (oligosaccharides containing about eight glucose units with one or more  $\alpha$ -1,6-glycosidic linkages), as well as smaller fragments such as **maltotriose** and **maltose**.



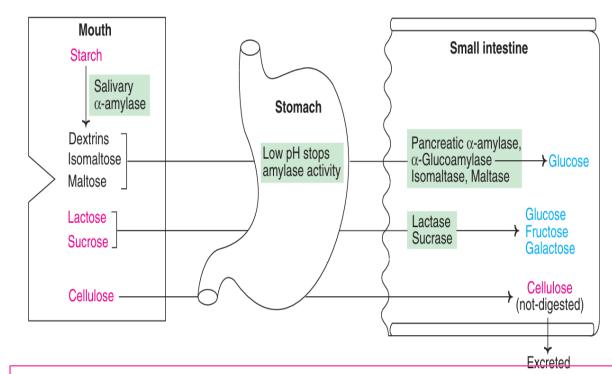
### **Digestion**

#### Carbohydrates are not digested in the stomach:

The enzyme salivary amylase is rapidly inactivated by the highly acidic environment (low pH) of the stomach. As a result, the hydrolysis of starch initiated in the mouth ceases, and no further enzymatic digestion of carbohydrates occurs in this compartment.

#### **Digestion in the Small Intestine**

- Pupon entry into the **small intestine**, the **acidic chyme** from the stomach is **neutralized** by **bicarbonate** secreted from the **pancreas**. **Pancreatic α-amylase** then continues starch hydrolysis by cleaving  $\alpha$ -1,4-glycosidic bonds, producing **disaccharides** (maltose, isomaltose) and **short oligosaccharides**.
- The final **breakdown** into **monosaccharides** occurs mainly at the **brush border** of the **upper jejunum**, through the action of oligosaccharidases (e.g., glucoamylase) and disaccharidases (e.g., maltase, sucrase, lactase).
- > Sucrase efficiently hydrolyzes **sucrose**, whereas lactase (β-galactosidase) acts more slowly, making **lactose** digestion a rate-limiting step in carbohydrate absorption in humans.



Note: Cellulose, hemicellulose, pectins, lignin and gums are not digested in the human digestive tract and constitute the main component of dietary fiber, playing an important role in intestinal transit and in the prevention of constipation.

#### **Absorption of Monosaccharides**

Carbohydrates are absorbed exclusively as **monosaccharides**. Because the enterocyte plasma membrane is **lipophilic**, monosaccharides cannot diffuse freely and require specific membrane transport proteins to facilitate their passage. Two major classes of carbohydrate transporters operate in **intestinal absorption**:

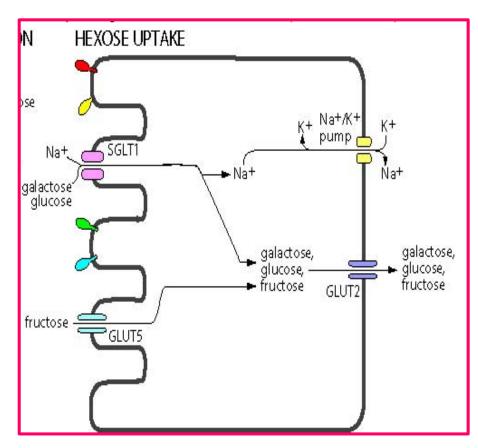
**SGLT** (Sodium–Glucose Linked Transporters): Mediate secondary active transport, driven by the sodium gradient maintained by ATP-dependent Na<sup>+</sup>/K<sup>+</sup>-ATPase.

**GLUT** (Glucose Transporters): Facilitate passive transport of monosaccharides by facilitated diffusion, without direct energy expenditure.

At the apical border of enterocytes, SGLT1 co-transports glucose, galactose, and Na<sup>+</sup> ions in the same direction across the membrane.

**GLUT5**, on the other hand, specifically mediates the facilitated diffusion of **fructose** into the cell.

At the basolateral membrane, GLUT2 facilitates the transport of glucose, fructose, and galactose. This membrane protein mediates the facilitated diffusion of these three monosaccharides from the enterocyte into the bloodstream, from where they are carried to the liver via the portal vein.



**Note:** Insulin increases the number and promotes the activity of **GLUT-4** in **skeletal muscle** and **adipose** tissue. In **type 2 diabetes mellitus**, insulin **resistance** is observed in these tissues. This is due to the **reduction** in the quantity of GLUT-4 in insulin deficiency.

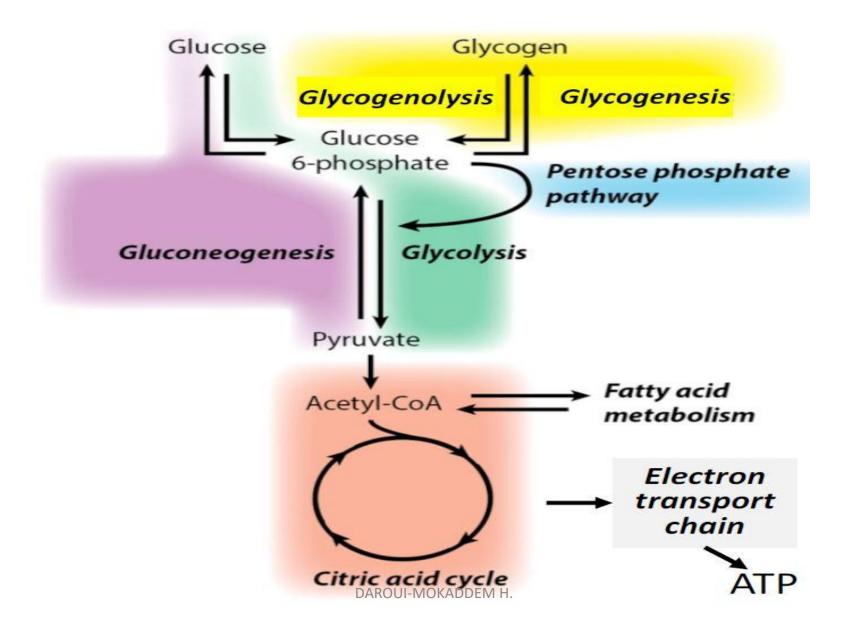
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#### Main Pathways of Carbohydrate Metabolism

The monosaccharides, glucose, galactose, and fructose resulting from the digestion and absorption of dietary carbohydrates are transported to the liver via the portal vein. Within hepatocytes, fructose and galactose are converted into glucose, which serves as the primary energy substrate of the body.

- Glucose in the liver can follow several metabolic fates:
- It may be **oxidized** to **pyruvate** through glycolysis, then enter the Krebs cycle to generate energy in the form of **ATP**.
- It may be stored as glycogen through the process of glycogenesis.
- Under conditions of excess glucose, when glycogen stores are saturated, it may be converted into fatty acids via lipogenesis.
- A small fraction (approximately **5–10**%) is metabolized through the pentose phosphate pathway, producing **NADPH**, which is essential for **lipogenic reactions**.
- The portion of glucose not taken up by the liver enters the systemic circulation via the hepatic veins. This circulating glucose represents a major energy source for the brain (which consumes about 50% of it), skeletal muscles (approximately 30%), and other peripheral tissues.
- During periods between meals or fasting, glucose is produced either from glycogen stores through glycogenolysis or by de novo synthesis from amino acids through gluconeogenesis.

### **Main Pathways of Carbohydrate Metabolism**



#### Carbohydrate Metabolism and Glucose Homeostasis

The primary function of carbohydrate metabolism is to maintain glucose homeostasis.

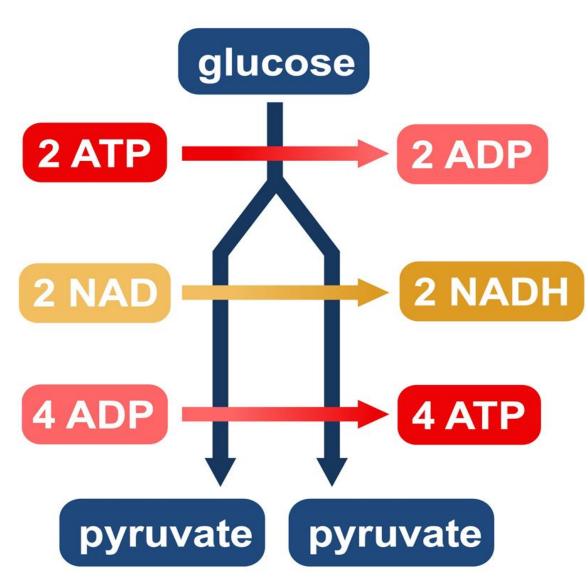
- ➤ Glucose homeostasis is a vital physiological process that ensures the maintenance of a stable blood glucose concentration (glycemia), thereby providing a constant supply of glucose to body organs.
- ➤ In a **healthy adult**, fasting blood glucos**e** levels range between 0.7 and 1.1 g/L, and remain below 1.4 g/L in the **postprandial state** (after a meal).
- This **regulation** is essential to prevent the adverse effects of fluctuating glucose supply to strictly glucose-dependent organs, such as the **brain**, **blood cells**, and **kidneys**.

*Note:* In diabetic individuals, these regulatory mechanisms are impaired, and glucose homeostasis is no longer maintained.

### Glycolysis pathway

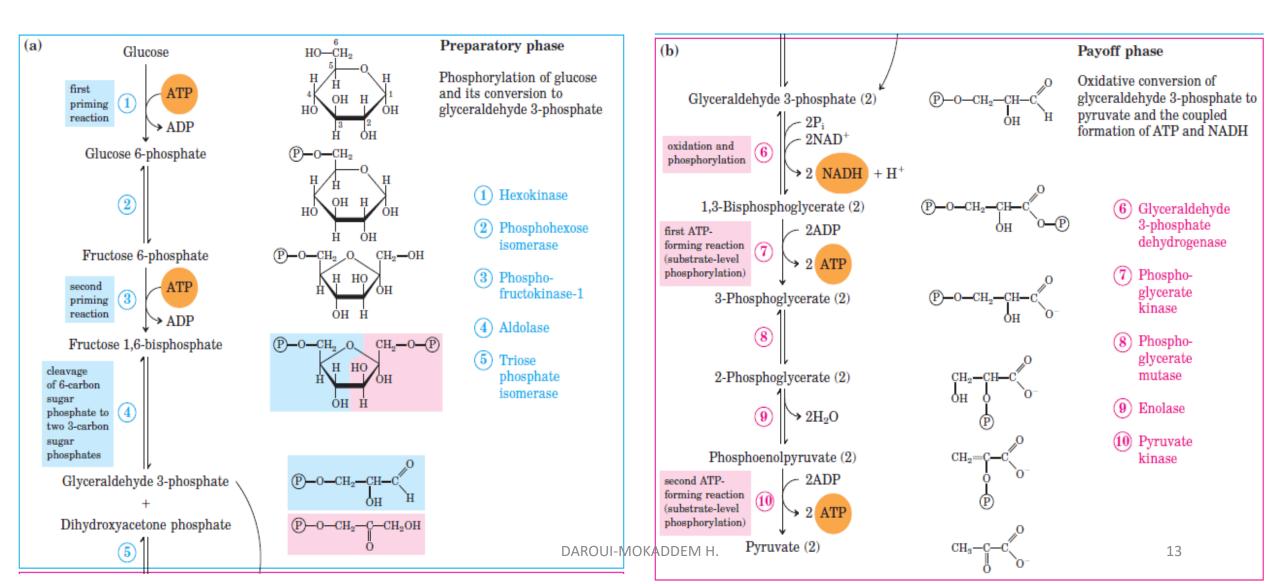
### Glycolysis pathway

- Glycolysis, also known as the Embden-Meyerhof pathway.
- The process takes place in the cytoplasm of a cell and does not require oxygen.
- It occurs in both aerobic and anaerobic organisms.
- Glycolysis is the process in which glucose is broken down to produce energy.
- It produces two molecules of pyruvate, ATP, NADH and water.



### glycolysis pathway

The **glycolytic pathway** consists of **ten enzymatic reactions**, all of which are reversible except for **three** key regulatory steps. These ten reactions are divided into two distinct phases: Preparatory phase (**five reactions**) and Payoffe phase (**five reactions**)



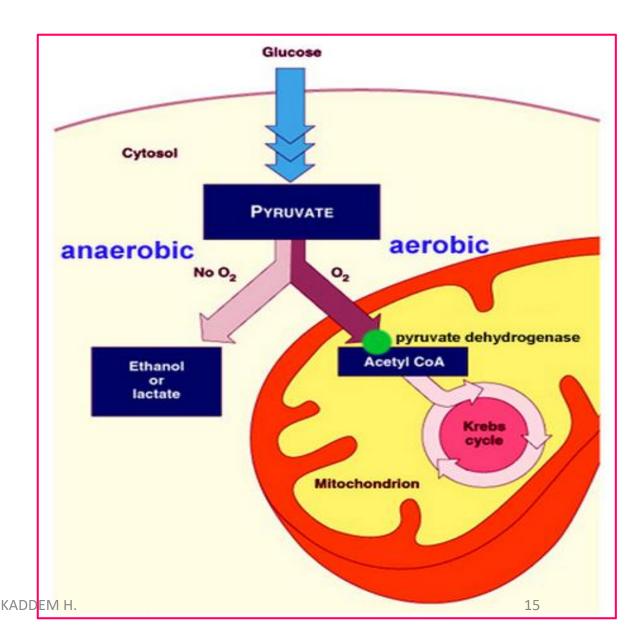
## The intermediates of glycolysis are phosphorylated compounds.

#### The phosphoryl group plays three essential roles:

- The **phosphate group** imparts a net **negative charge** to all intermediates between **glucose** and **pyruvate** at physiological pH (pH 7), thereby preventing their diffusion out of the cell.
- > The phosphate group participates in **energy conservation**, enabling the generation of **two molecules** of **ATP** during glycolysis.
- > The phosphate group also functions as a binding and recognition element, facilitating the formation of enzyme- substrate complexes.

### Position of Glycolysis in Energy Metabolism

- It involves the degradation of glucose with the production of ATP and intermediate metabolites, which can be further utilized in other metabolic pathways.
- Aerobic glycolysis yields 2 ATP, 2 NADH + H<sup>+</sup>, and 2 pyruvate molecules.
- leading to the formation of **pyruvate** and its **mitochondrial** conversion into acetyl-CoA, the key substrate fueling the Krebs cycle (tricarboxylic acid cycle). Under these conditions, glucose undergoes **complete oxidation** to **CO<sub>2</sub>** and **H<sub>2</sub>O.**

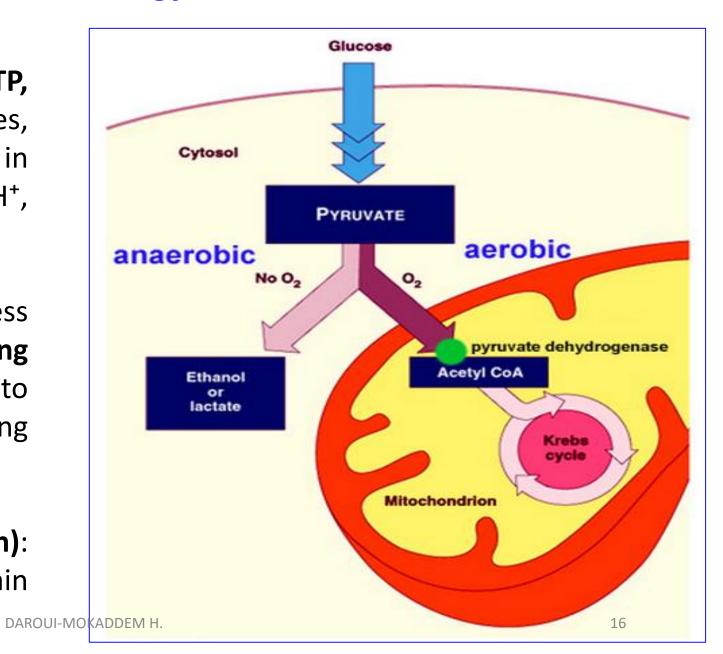


### Position of Glycolysis in Energy Metabolism

Anaerobic glycolysis also produces 2 ATP, 2 NADH + H<sup>+</sup>, and 2 pyruvate molecules, which are subsequently utilized in reactions that reoxidize NADH + H<sup>+</sup>, leading to the formation of:

Lactate (lactic fermentation): a process that provides ATP to cells lacking mitochondria (e.g., red blood cells) and to hypoxic tissues such as rapidly contracting skeletal muscle.

Ethanol (alcoholic fermentation): occurring in yeasts and certain microorganisms.



### Cytoplasmic Glycolysis Balance

#### The general reaction is written as:

```
1 glucose (C_6H_{12}O_6) + 2 \text{ ADP} + 2 \text{ Pi} + 2 \text{ NAD}^+ 2 pyruvates (CH3-CO-COO^-) + 2 \text{ ATP} + 2 H_2O + 2 \text{ NADH}, H^+
```

#### **Energy balance under anaerobic conditions (up to pyruvate):**

Two reactions consume energy: – 2 ATP

Two reactions generate energy: + 4 ATP

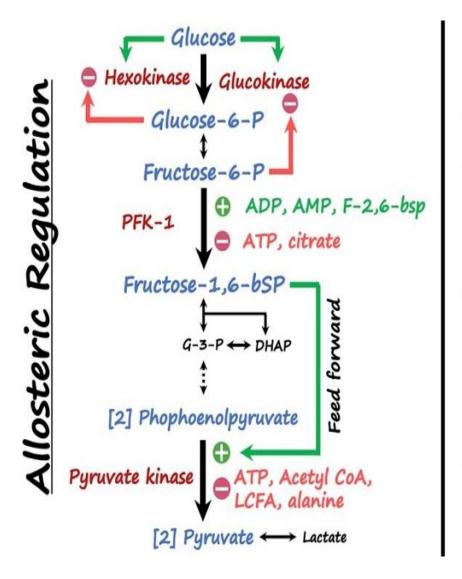
The net gain is therefore 2 ATP.

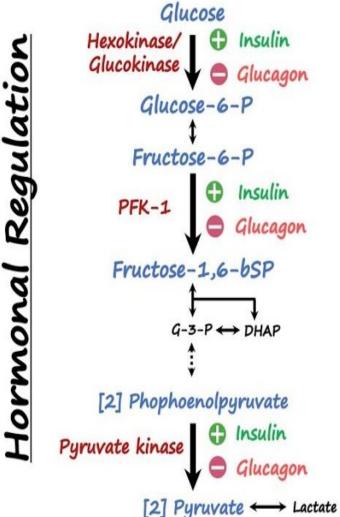
#### **Energy balance under aerobic conditions:**

The 2 NADH + H<sup>+</sup> produced are oxidized through the respiratory chain, yielding  $3 \text{ ATP} \times 2 = 6 \text{ ATP}$ .

Under **aerobic conditions**, the total energy yield is 2 + 6 = 8 ATP.

### **Regulation of Glycolysis**





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#### **Hormonal Regulation**

➤In the **postprandial state** (after a meal), **insulin** increases hepatic levels of **glucokinase**, **phosphofructokinase-1** (PFK-1), and **pyruvate kinase**.

These changes result from the activation of gene transcription for the corresponding enzymes.

The **increase** in these enzyme concentrations enhances glycolytic activity.

Conversely, during fasting, when blood glucagon levels are elevated, the transcription of these genes is downregulated, leading to a decrease in the concentrations of these enzymes.

Glucagon **inhibits** glycolysis, thereby preserving glucose for **glucose**-**dependent tissues** such as the **brain** and **kidneys**.

### Allosteric regulation of Glycolysis

Purpose: To adjust the rate of glycolysis according to the cell's metabolic demands:

- For energy production (ATP)
- For the supply of biosynthetic precursors

#### The rate of glycolysis depends on:

- The cellular availability of glucose
- The activity of the rate-limiting reactions (steps 1, 3, and 10) catalyzed respectively by hexokinase (or glucokinase), phosphofructokinase-1 (PFK-1), and pyruvate kinase.
- The major regulatory step is **reaction 3**, catalyzed by **PFK-1**, which irreversibly commits glucose to the glycolytic pathway.

Hexokinase: Slightly limiting step. Allosterically inhibited by its product, glucose-6-phosphate (G6P).

*Phosphofructokinase-1* (PFK-1): **Key regulatory enzyme** and the most rate-limiting step of glycolysis. **Allosterically** regulated by several **activators** and **inhibitors**.

<u>Activators</u>: AMP, ADP, **Fructose-2,6-bisphosphate**, Fructose-6-phosphate

<u>Inhibitors</u>: ATP, Citrate

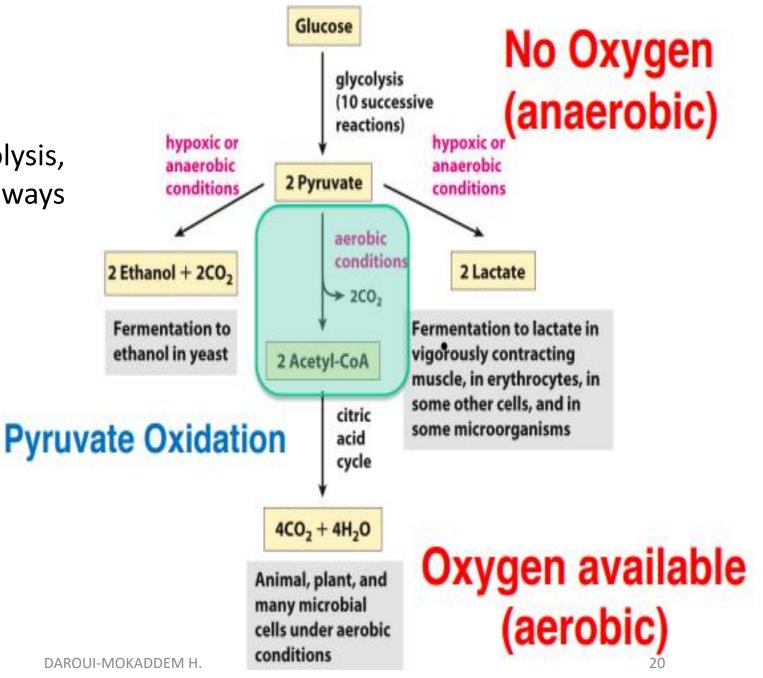
Pyruvate Kinase: Allosterically inhibited by ATP.

**Note:** Fructose-2,6-bisphosphate is derived, like fructose-1,6-bisphosphate, from fructose-6-phosphate, but it functions solely as a regulatory molecule. Its synthesis and degradation are under hormonal control.

### **Fates of Pyruvate**

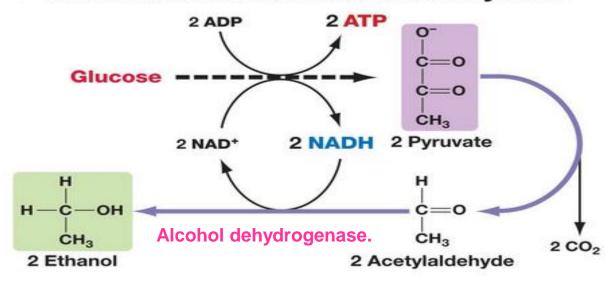
Pyruvate, the end product of glycolysis, can follow different catabolic pathways depending on:

- The type of organism
- The metabolic conditions.



#### Ethanol pathway

#### Alcohol fermentation occurs in yeast.

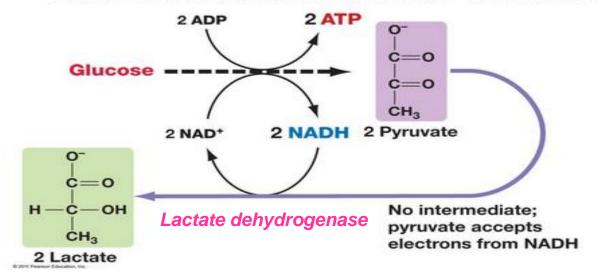


#### **Alcoholic Fermentation:**

In yeasts and other microorganisms, **NADH** is **used** to **reduce** pyruvate to ethanol through the sequential action of the enzymes pyruvate decarboxylase and *alcohol dehydrogenase*.

#### Lactate pathway

#### Lactic acid fermentation occurs in humans.



#### **Lactic Fermentation:**

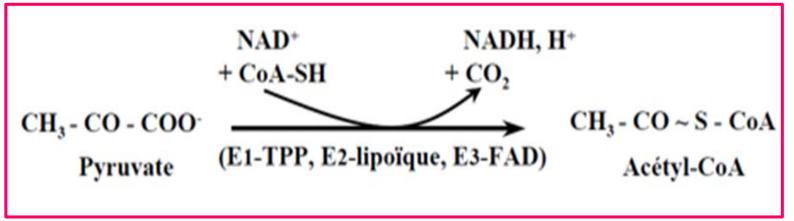
Under anaerobic conditions, the NADH produced during glycolysis cannot enter the mitochondria and therefore does not participate in the respiratory chain.

Instead, it is u**sed** to r**educe** pyruvate to lactate, a reaction catalyzed by the enzyme *lactate dehydrogenase*.

#### This process occurs in:

- Intensely contracting muscles
- Red blood cells
- Certain microorganisms.

Oxidative decarboxylation of pyruvate into acetyl-CoA (Aerobiosis)



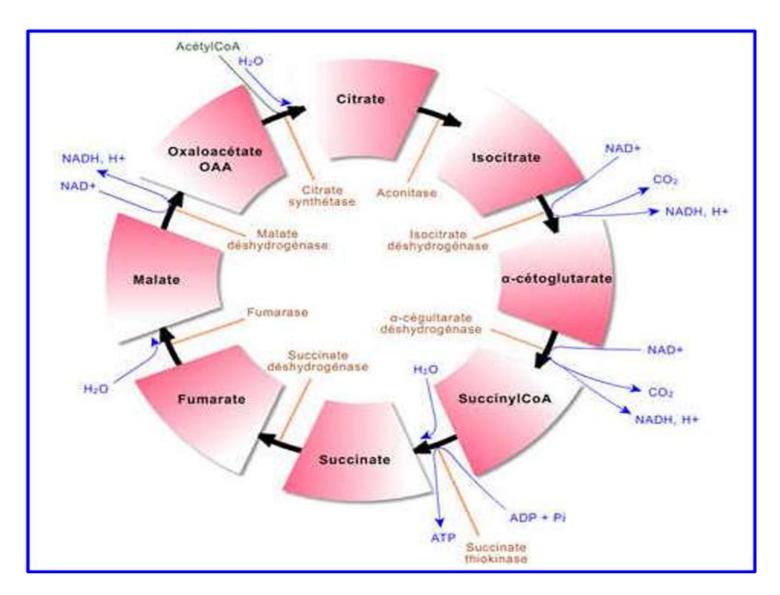


In the **mitochondrion**, the **oxidative decarboxylation** of pyruvate into acetyl-CoA is catalyzed by *pyruvate* dehydrogenase (PDH)

**Required Cofactors:** Thiamine pyrophosphate (**TPP**), Lipoic acid (lipoamide), Coenzyme A (CoA-SH), FAD (flavin adenine dinucleotide), NAD<sup>+</sup> (nicotinamide adenine dinucleotide).

- This reaction is **highly exergonic** (irreversible). It leads to the formation of a high-energy **thioester** bond in acetyl-CoA and the production of **NADH,H**<sup>+</sup>, which will yield **3 ATP** molecules in the **respiratory chain**.
- > Pyruvate dehydrogenase is regulated by its substrates:
- increased [pyruvate], NAD+, CoA-SH, and AMP (activators);
- NADH, acetyl-CoA, and ATP (inhibitors).

### Krebs cycle (Tricarboxylic acid cycle)

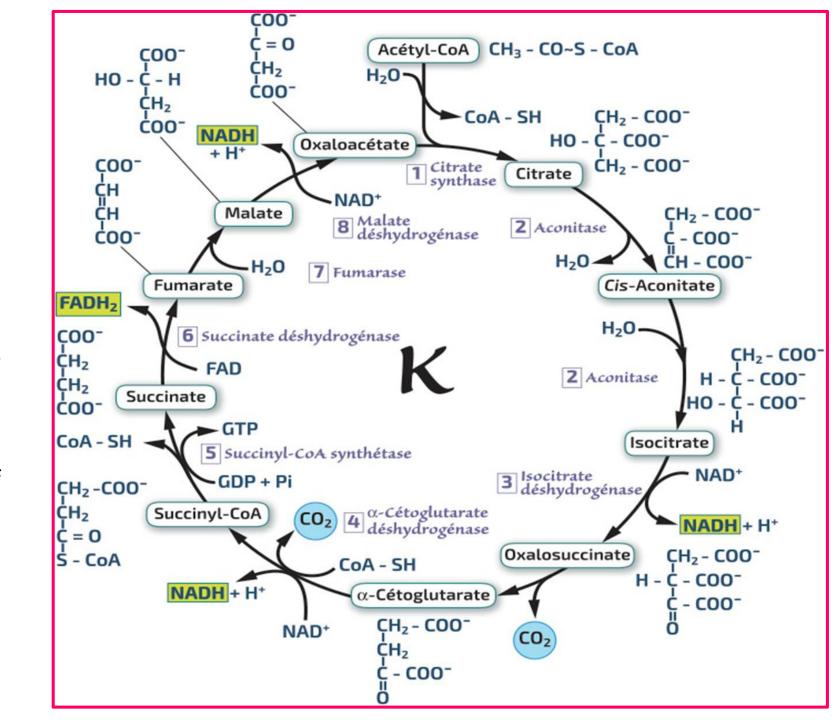


### Krebs cycle

#### **Metabolic Importance:**

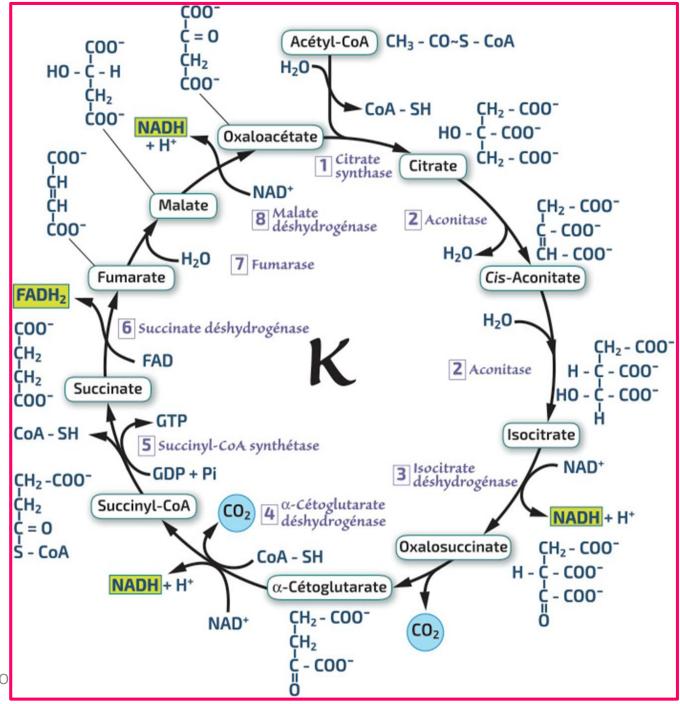
The Krebs cycle is **amphibolic**, meaning that it plays a dual role:

- Catabolic, by oxidizing acetyl CoA to produce energy.
- Anabolic, by providing metabolic intermediates for the biosynthesis of amino acids, purine bases, heme, and glucose (via gluconeogenesis).



### Krebs cycle

- The Krebs cycle, or citric acid cycle, takes place in the **mitochondrion** in **eukaryotic cells**.
- At each turn of the cycle, one molecule of acetyl-CoA (2 carbons), derived from carbohydrates, condenses with one molecule of oxaloacetate (4 carbons) to form citrate (6 carbons).
- The cycle consists of eight reactions catalyzed by **seven soluble enzymes** and one enzyme anchored in the inner mitochondrial membrane: *succinate dehydrogenase*.
- ➤ It generates **reduced coenzymes** (3 NADH + H<sup>+</sup> and 1 FADH<sub>2</sub>), which will be reoxidized in the **respiratory chain**. Therefore, the **Krebs cycle** is functionally **coupled** to the respiratory chain.
- ➤ During the sequence of reactions, **two carbons** of citrate are **released** as **CO<sub>2</sub>**, thereby allowing the regeneration of oxaloacetate.



### Energy balance of the Krebs cycle

The overall reaction of the cycle

Acetyl-CoA + 
$$3$$
NAD<sup>+</sup> + FAD + GDP+P<sub>i</sub> +  $2$ H<sub>2</sub>O  $\longrightarrow$   $2$ CO<sub>2</sub> +  $3$ (NADH, H<sup>+</sup>) + FADH<sub>2</sub> + GTP + COASH

- Each turn produces
- 1 ATP
- 3 (NADH,  $H^+$ ) = 3 X 3 ATP = 9 ATP
- 1 FADH2 = 2 ATP
- In total, one molecule of acetyl-CoA produces 12 ATP per cycle turn.

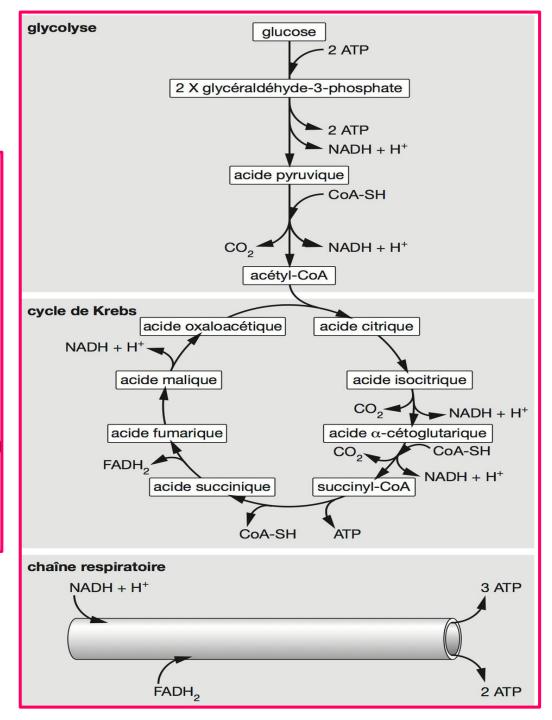
### Regulation du cycle de Krebs

- > The cycle is **regulated** by the availability of its **substrates** and by the cell's **energetic state**:
- Inhibited by ATP, NADH, and succinyl-CoA (high-energy state).
- Activated by ADP and Ca<sup>2+</sup> (increased energy demand).
- > The main regulated enzymes are:
- Citrate synthase
- Isocitrate dehydrogenase
- α-Ketoglutarate dehydrogenase.

### Total energy yield from glucose degradation

	Réaction	ATP ou Coenzymes réduits formés □□ATP				
Glucose	<b></b>	Glucose 6P	- 1 ATP	- 1		
Fructose 6P	<b>→</b>	Fructose 1,6 bisP	- 1 ATP	- 1		
2 Glycéraldéhyde 3P	$\rightarrow$	2 1,3 bis Phospho Glycérate	2 NADH	6	<b>\&gt;8</b>	
2 1,3 bis Phospho Glycérate	<b>→</b>	2 3 Phospho Glycérate	2 ATP	2		
2 Phosphoénolpyruvate	$\rightarrow$	2 Pyruvate	2 ATP	2	J	
2 Pyruvate	<b>†</b>	2 AcétylCoA	2 NADH	6	6	
2 Isocitrate	<b>—</b>	2 α cétoglutarate	2 NADH	6		
2 α cétoglutarate	<b>→</b>	2 SuccinylCoA	2 NADH	6		
2 SuccinylCoA	<b>→</b>	2 Succinate	2 GTP	2	<b>≥2</b>	
2 Succinate	<del></del>	2 Fumarate	2 FADH <sub>2</sub>	4		
2 L-malate	<b>—</b>	2 Oxaloacétate	2 NADH	6		
Total ATP formés : 38						

NADH,H<sup>+</sup> and FADH<sub>2</sub> are reoxidized by the respiratory chain.



### Clinical aspect of glycolysis

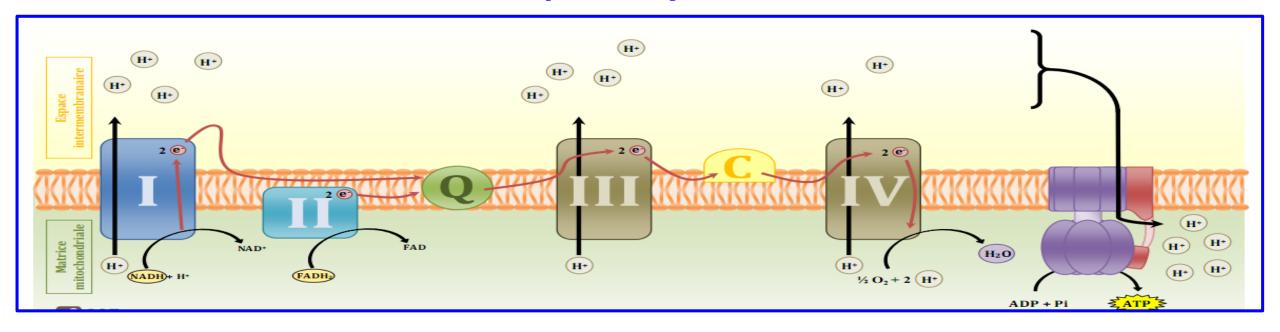
Clinical Aspect	Description	Consequences / Manifestations	
Tissues Dependent on Glycolysis	Red blood cells (lack mitochondria), brain, muscle under anaerobic conditions	Muscle fatigue, increased sensitivity to hypoglycemia, hemolysis if glycolytic enzymes are defective	
Cancer (Warburg Effect)	Increased glycolytic flux even in the presence of oxygen	High glucose uptake, basis for PET imaging using 18-FDG	
Lactic Acidosis	Excess lactate production or impaired lactate clearance	Decreased blood pH, respiratory distress, medical emergency	
Enzymatic Deficiencies	Pyruvate kinase, phosphofructokinase (Tarui disease), hexokinase defects	Hemolytic anemia, exercise intolerance, myoglobinuria	
Intense Exercise	Anaerobic glycolysis leading to elevated lactate	Muscle cramps, decreased pH, fatigue	
Hypoglycemia	Insufficient glucose supply to the brain	Neurological symptoms: confusion, seizures, coma	
Diabetes Mellitus	Chronic hyperglycemia causing metabolic overload	Vascular complications, increased oxidative stress	

### Clinical aspect of Krebs cycle

- Mutations affecting Krebs cycle enzymes have profound clinical and metabolic effects. They can cause severe metabolic encephalopathies in infants, as well as a broad spectrum of tumors and cancers in adults.
- The underlying mechanisms include impaired **mitochondrial oxidation**, the accumulation of **oncometabolites**, and extensive **epigenetic modifications** of **DNA** and **histones**.

### Respiratory chain

### Respiratory chain



- The respiratory chain is located in the inner mitochondrial membrane. This electron transport chain consists of four protein complexes:
- Complex I: NADH-coenzyme Q oxidoreductase,
- Complex II: Succinate-coenzyme Q oxidoreductase,
- Complex III: Coenzyme Q-cytochrome c oxidoreductase,
- Complex IV: Cytochrome c oxidase.
- Coenzyme Q (ubiquinone) and cytochrome c are mobile electron carriers within the respiratory chain.

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### Respiratory chain

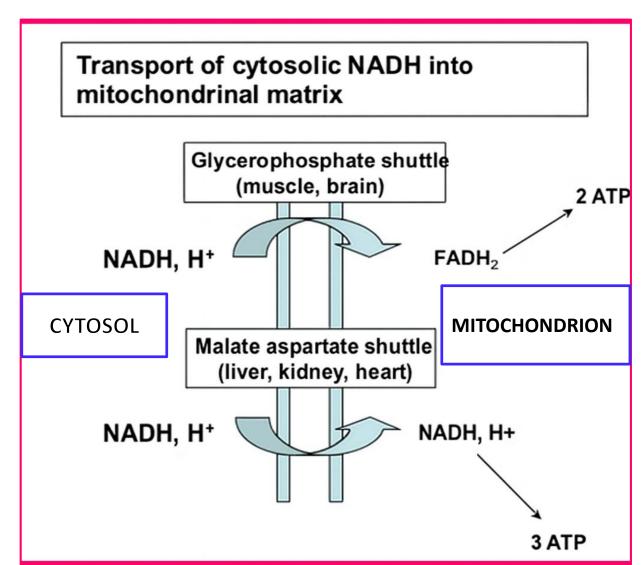
It produces **ATP** and **water** through a **coupled process** composed of two main components:

- ➤ The **electron transport chain** generates H<sub>2</sub>O by transferring protons (H<sup>+</sup>) from the reduced coenzymes NADH,H<sup>+</sup> and FADH<sub>2</sub> to molecular oxygen, this process constitutes **cellular respiration**.
- The **phosphorylation of ADP into ATP** is driven by the energy released from the electron transport chain. The coupling of these two processes is known as **oxidative phosphorylation**.
- During this process, ADP is phosphorylated to ATP by the enzyme ATP synthase, yielding approximately 3 ATP per NADH,H<sup>+</sup> and 2 ATP per FADH₂.
- > The **ADP** and inorganic phosphate (**Pi**) used for ATP synthesis originate from the cytosol.
- The control of the **respiratory chain** and **ATP synthesis** depends on the **concentration** of **ADP**. When ADP concentration increases, the **rate** of the respiratory chain **rises rapidly** and **significantly**.

### Clinical aspect of the mitochondrial respiratory chain

- > The mitochondrial respiratory chain produces the energy required by our cells.
- ➤ When it is impaired, high-energy-demand organs such as the **brain** and **muscles** are particularly affected.
- > Symptoms include **fatigue**, **muscle weakness**, and **neurological disorders**. Certain genetic diseases, known as **mitochondrial myopathies**, are associated with these defects.

# Shuttle system 36 or 38 ATP! Depending on the shuttle system used, one molecule of glucose will generate 36 or 38 / molecules of ATP.



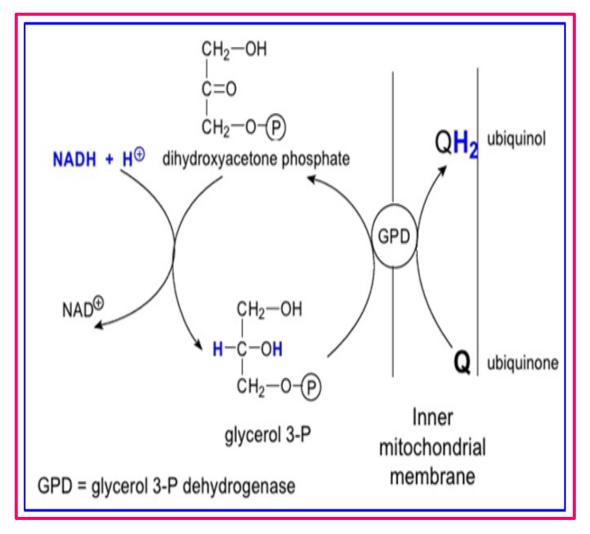
#### Shuttle system

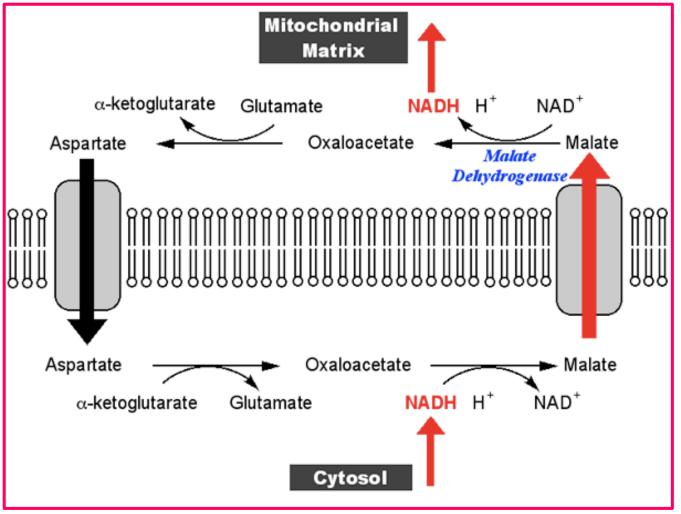
#### (Transport of cytosolic NADH to the mitochondrial matrix)

- ➤ During **glycolysis**, the **cytosolic NADH** produced in step 6 catalyzed by **glyceraldehyde-3- phosphate dehydrogenase** must be regenerated into NAD<sup>+</sup>, otherwise glycolysis would stop due to insufficient NAD<sup>+</sup>.
- The problem: The mitochondrial membrane is **impermeable** to cytosolic NADH.
- The solution: Eukaryotic cells possess shuttle systems that transfer the electrons from cytosolic NADH into the mitochondria without the NADH molecule itself crossing the inner membrane.
- Depending on the tissue, there are two types of shuttles:
- Glycerol phosphate shuttle (muscle, brain)
- Malate—aspartate shuttle (liver, kidneys, heart)

#### The glycerol phosphate shuttle

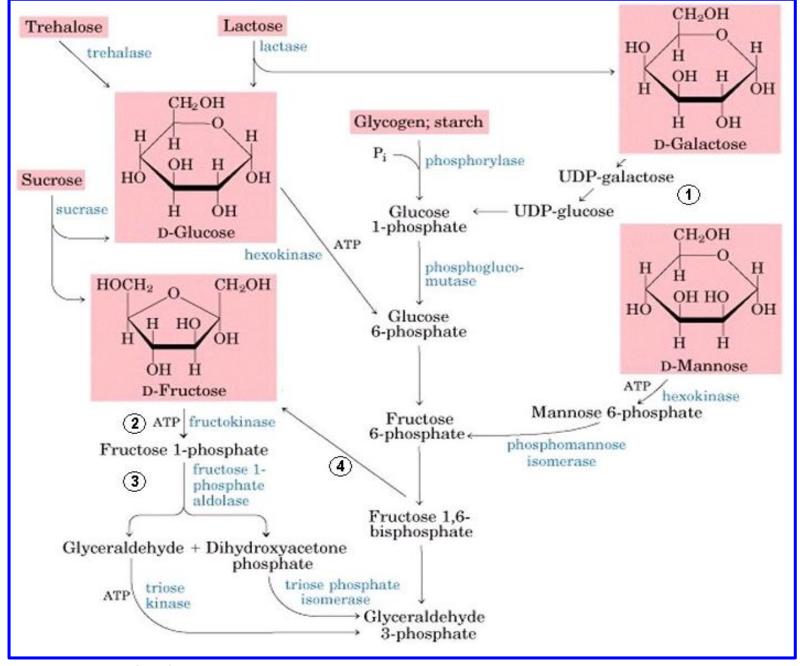
#### The malate-aspartate shuttle





# Gluconeogenesis from other hexoses

- Fructose, galactose, and mannose are required for the biosynthesis of glycoproteins.
- They can also serve as energy substrates.
- Their conversion into glucose occurs in the liver. Indeed, these hexoses undergo a series of parallel reactions that allow them to enter the glycolytic pathway at an appropriate intermediate step.



#### **Fructose Metabolism**

## **Clinical aspect:**

➤ Hepatic *fructokinase* deficiency: This results in essential fructosuria, which is generally benign and asymptomatic.

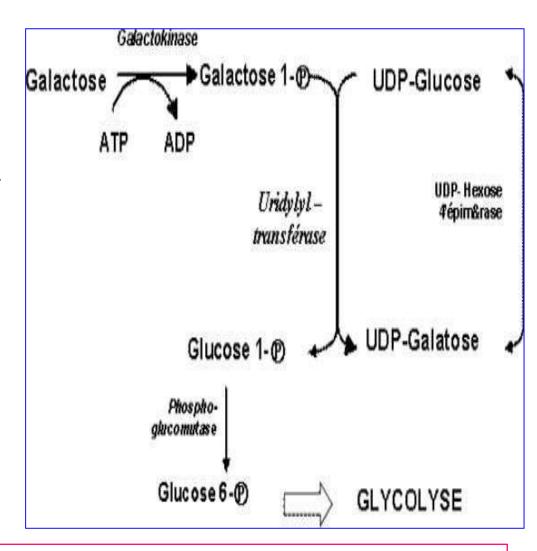
➤ Aldolase B deficiency: This leads to hereditary fructose intolerance, characterized by the accumulation of fructose-1-phosphate, leading to hypoglycemia, nausea, vomiting, and potential liver damage.

#### **Galactose Metabolism**

#### **Clinical aspect:**

Galactose-1-phosphate uridyltransferase deficiency causes congenital galactosemia, the most common inherited disorder of carbohydrate metabolism.

- ➤ Galactosemia is characterized by the accumulation of galactose in the blood and tissues, leading, among other effects, to severe hypoglycemia and early-onset cataracts.
- ➤ Galactose is **reduced** by *aldose reductase* to galactitol, which opacifies the lens.
- Symptoms of this disease include **vomiting**, **diarrhea**, and **mental retardation**; a galactose-free diet prevents the manifestation of the disease. If needed, the body can synthesize **galactose** from **glucose**.



**Note**: The same enzyme, aldose reductase, reduces glucose to sorbitol, which is responsible for cataracts in diabetics and can contribute to liver dysfunction.

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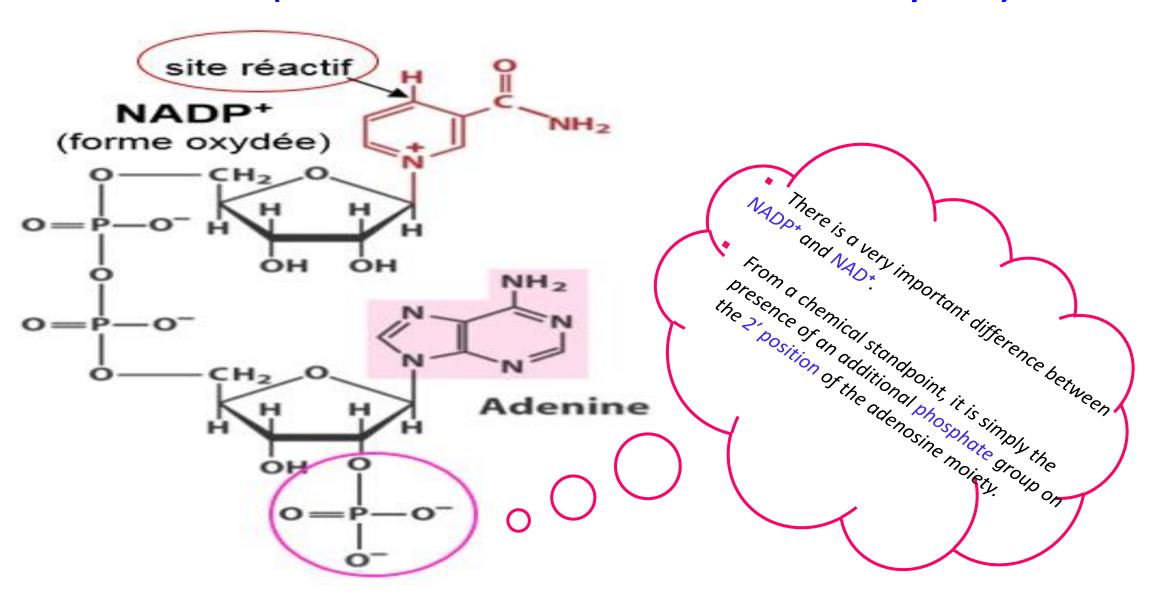
# Pentose Phosphate Pathway (PPP)

#### Pentose Phosphate Pathway (PPP)

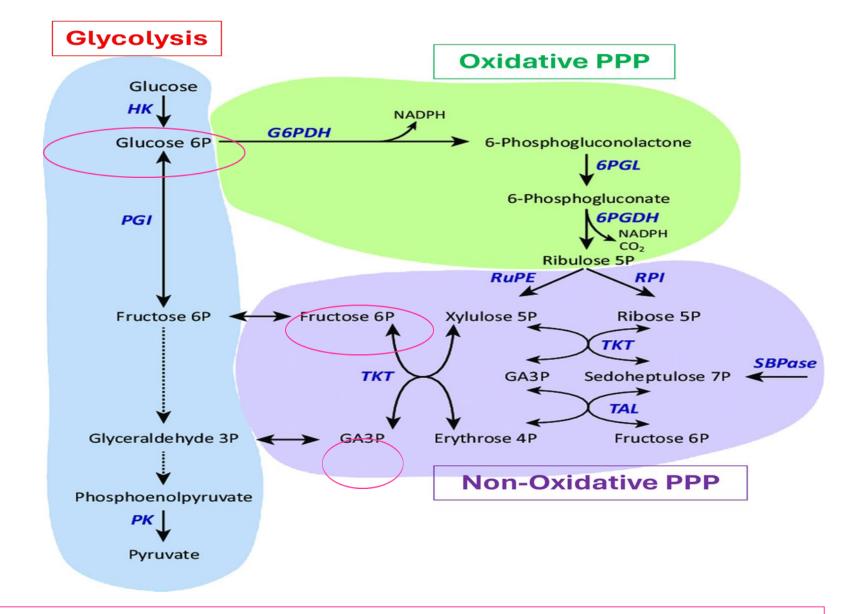
- The PPP occurs in the cytosol.
- ➤ Unlike **glycolysis**, this pathway does not generate energy.
- Its primary function is the production of:
- NADPH, H<sup>+</sup>: This molecule plays a critical role in the **detoxification** of peroxides in **red blood cells**, as well as in the **biosynthesis** of **fatty acids**, **cholesterol**, and **steroid hormones**. NADPH, H<sup>+</sup> provides the reducing power required for numerous cellular reactions.
- Ribose-5-phosphate and its derivatives: These compounds are essential for the **synthesis** of nucleic acids (**RNA** and **DNA**) and nucleotides (such as **ATP**, **coenzyme A**, **NAD**, **FAD**, etc.). The pathway also produces erythrose-4-phosphate, which serves as a precursor for the biosynthesis of **aromatic amino** acids.

**Note**: The pentose phosphate pathway is **ubiquitous** (it operates in all cells), but its **activity** varies according to the cellular requirements for **ribose** and **NADPH**,  $H^+$ .

# Chemical structure of NADP<sup>+</sup> (Nicotinamide Adenine Dinucleotide Phosphate)



# Overview of the Pentose Phosphate Pathway

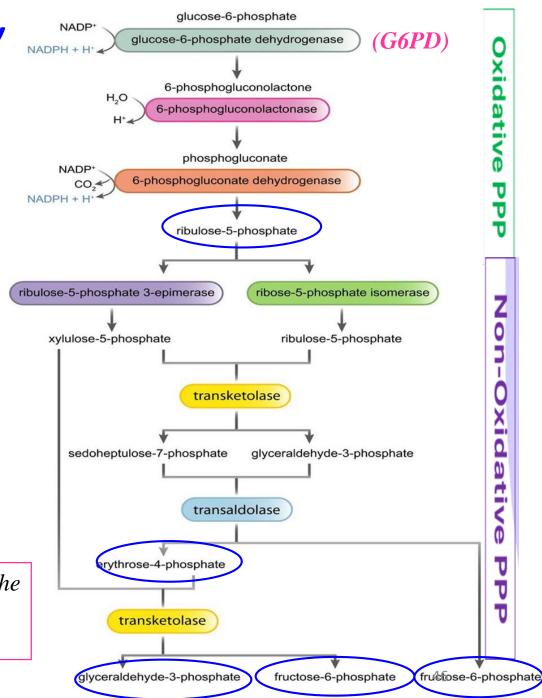


**Note:** The starting substrate is **glucose-6-phosphate**, and the final products are **fructose-6-phosphate** and **glyceraldehyde-3-phosphate**; thus, both the **initial** and **final** metabolites of the pathway are intermediates of **glycolysis**.

## The phases of the pentose phosphate pathway

- The ribose-5-phosphate produced is utilized for the synthesis of **nucleic acids** or other nucleotides.
- ➤ In **cells** that require **only** NADPH, H<sup>+</sup>, **ribose-5-phosphate** is redirected back into **glycolysis**.
- Erythrose-4-phosphate is used for the biosynthesis of aromatic amino acids (tyrosine, tryptophan, and phenylalanine), or it can be converted together with xylulose-5-phosphate into glyceraldehyde-3-phosphate and fructose-6-phosphate.
- The intermediates generated (glyceraldehyde-3-phosphate and fructose-6-phosphate) can eventually enter the glycolytic pathway, serving as connection points between the pentose phosphate pathway and glycolysis.

**Note:** Glucose-6-phosphate dehydrogenase (G6PD) is the **key enzyme** of the pentose phosphate pathway. The reactions of the **non-oxidative phase** are reversible and are primarily **regulated** by substrate availability.

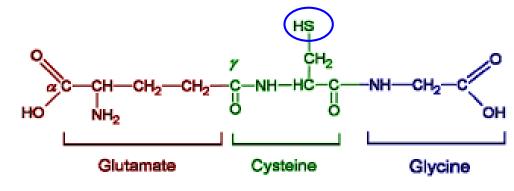


## Role of NADPH and Glutathione in Red Blood Cells

The pentose phosphate pathway supplies red blood cells with NADPH, which is required to recycle oxidized glutathione (GS–SG) back to its reduced form (GSH).

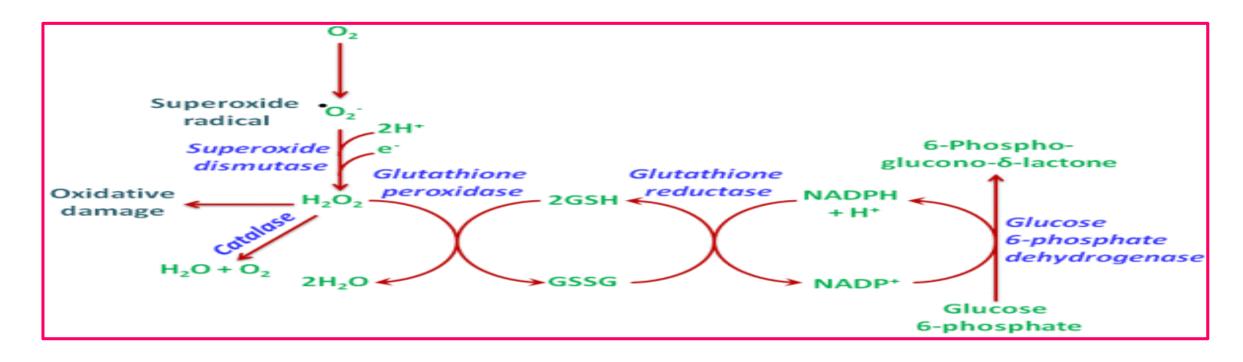
ightharpoonup Reduced glutathione protects red blood cells from damage caused by toxic oxidizing molecules, such as hydrogen peroxide ( $H_2O_2$ ).

#### **GLUTATHIONE (GSH)**



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# Role of NADPH and Glutathione in Red Blood Cells



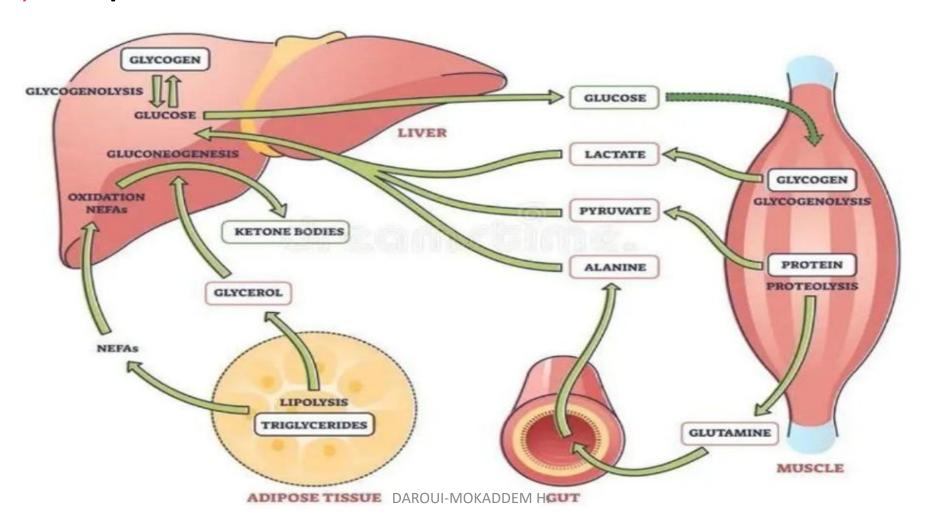
A deficiency in glucose-6-phosphate dehydrogenase (G6PD) decreases the efficiency of the pentose phosphate pathway in producing NADPH. This reduction leads to the accumulation of organic peroxides, which deform the red blood cell membrane and oxidize hemoglobin (Fe<sup>2+</sup>  $\rightarrow$  Fe<sup>3+</sup>) as well as other erythrocyte proteins.

**Note**: The main clinical manifestation is an acute hemolytic crisis occurring a few hours after ingestion of an oxidizing agent (e.g., **fava beans**, **drugs** such as **aspirin** or **sulfonamides**), resulting in **hemolytic anemiaa** condition known as **favism**.

# Gluconeogenesis

# **Gluconeogenesis**

**Definition**: the formation of **carbohydrate molecules**, primarily **glucose** and **glycogen**, from non-carbohydrate **precursors**.



# Gluconeogenesis

- Certain tissues (the brain, red blood cells, rapidly contracting muscle, etc.) require a continuous supply of glucose. The liver can fulfill this function by mobilizing glycogen (short-term) and through gluconeogenesis.
- > In humans, gluconeogenesis occurs primarily in:
- The liver: 90% of newly synthesized glucose
- The kidneys and intestinal epithelium: 10%
- It does not occur in muscle or brain tissue.

**Note:** The reactions of gluconeogenesis are **c**onserved across animals, **plants**, **fungi**, and **microorganisms**.

## **Precursors of gluconeogenesis**

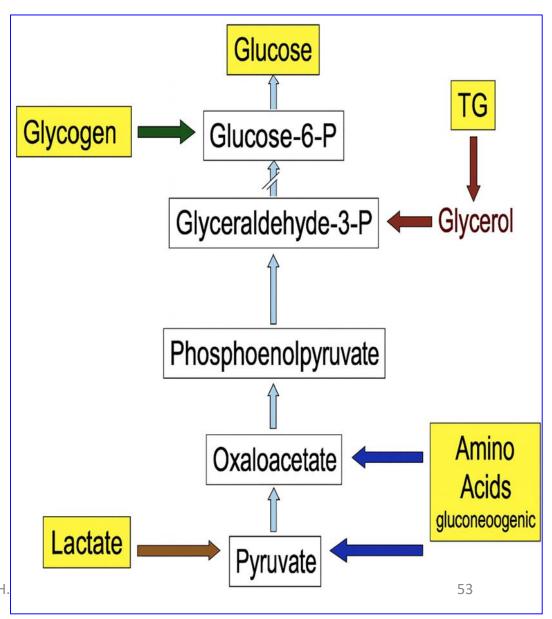
- Pyruvate and lactate (≈ 1/3): derived from red blood cells and muscle cells.
- ightharpoonup Alanine ( $\approx 1/3$ ): an amino acid originating from muscle cells.
- Glycerol: derived from the catabolism of dietary triglycerides, adipose tissue, or circulating lipoproteins.
- > Gluconeogenic amino acids: obtained from dietary proteins or tissue proteins.
- Propionate: derived from the degradation of odd-chain fatty acids.

**Note:** In humans and animals, **acetyl-CoA** (produced from fatty acid degradation) cannot be converted into **glucose**; therefore, **fatty acids** cannot serve as substrates for **gluconeogenesis**.

## **Entry of Precursors into Gluconeogenesis**

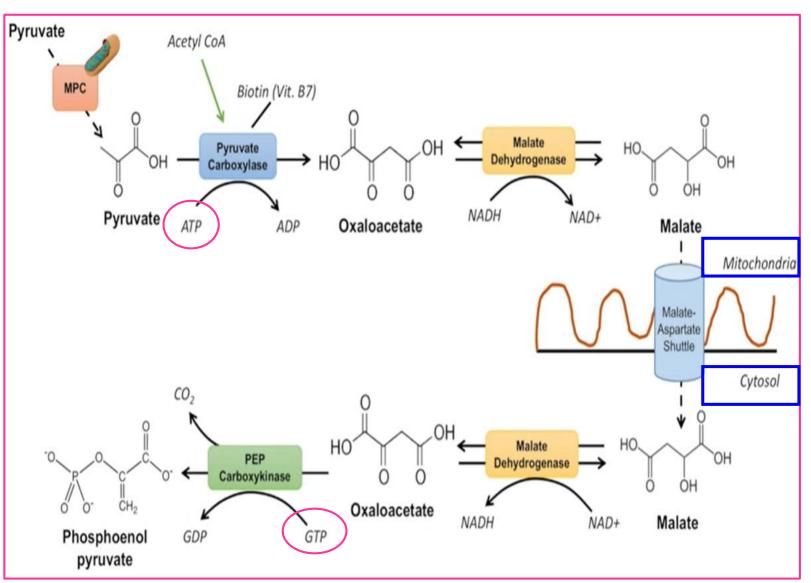
The three entry points of precursors into gluconeogenesis are:

- Pyruvate: for lactate, alanine, and gluconeogenic amino acids whose catabolism converges at pyruvate.
- Phosphoenolpyruvate (PEP): for gluconeogenic amino acids whose catabolism feeds into intermediates of the Krebs cycle.
- Dihydroxyacetone phosphate (DHAP): for glycerol.



#### Gluconeogenesis from Pyruvate

- Pyruvate is transported into the mitochondrial matrix and then carboxylated to oxaloacetate by pyruvate carboxylase, a strictly mitochondrial enzyme requiring the coenzyme biotin (vitamin B8). This reaction consumes one molecule of ATP.
- The phosphorylative decarboxylation of oxaloacetate to phosphoenolpyruvate (PEP) consumes one molecule of GTP (regenerated from ATP) and is catalyzed by phosphoenolpyruvate carboxykinase (PEPCK).



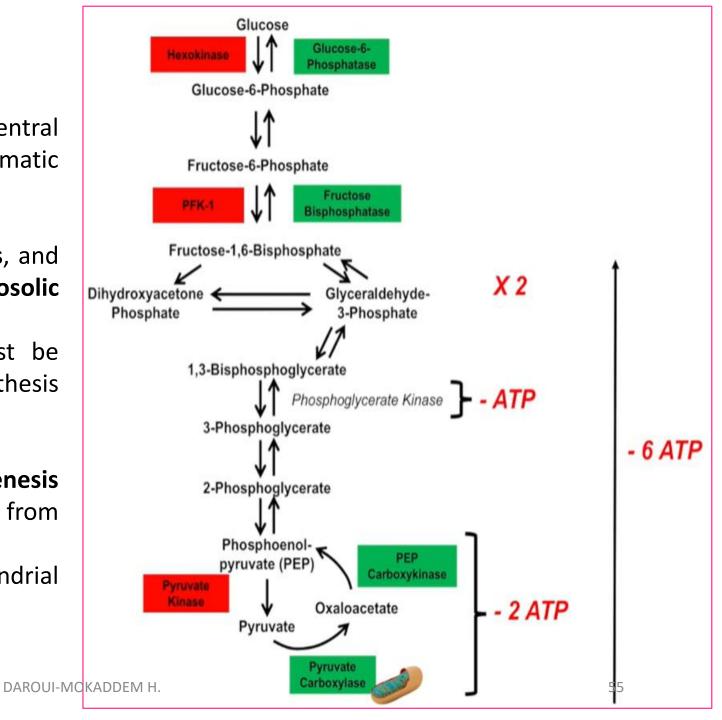
#### **Derivation of Gluconeogenesis**

The **conversion** of **pyruvate** into **glucose** is the central pathway of gluconeogenesis. Among its ten enzymatic reactions:

- > Seven are the reverse reactions of glycolysis, and are therefore catalyzed by the same cytosolic enzymes.
- Three **irreversible** glycolytic reactions must be bypassed in gluconeogenesis so that glucose synthesis becomes thermodynamically favorable.
- Thus, reactions 1, 8, and 10 of **gluconeogenesis** are catalyzed by enzymes that are different from those of glycolysis:

*Pyruvate carboxylase*, which are mitochondrial enzymes,

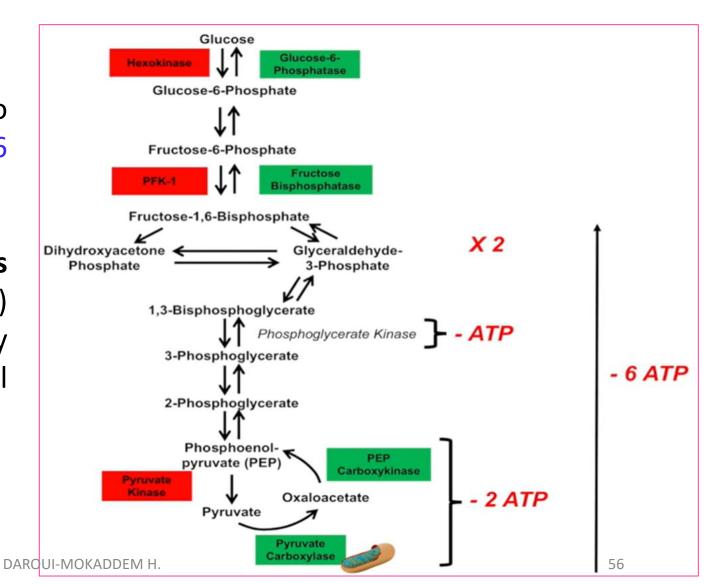
Fructose-1,6-bisphosphatase (F1,6BPase), and Glucose-6-phosphatase.



## **Energetic Balance of Gluconeogenesis**

2 pyruvates + 4 ATP + 2 GTP + 2 (NADH, H<sup>+</sup>) + 6 H2O - Glucose + 4 ADP + 2 GDP + 6 Pi + 2 NAD<sup>+</sup> + 2H+

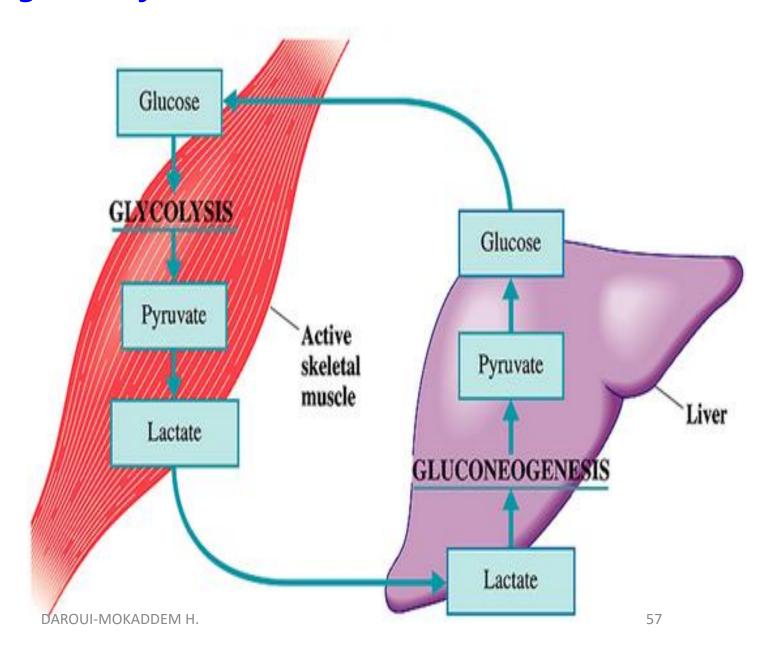
- From **two molecules** of **pyruvate** to one molecule of glucose, 2 × 3 ATP = 6 ATP are consumed.
- ➤ The energetic cost of gluconeogenesis versus glycolysis (6 ATP / 2 ATP) reflects the price of the irreversibility of these pathways and their reciprocal regulation.



## Gluconeogenesis from Lactate

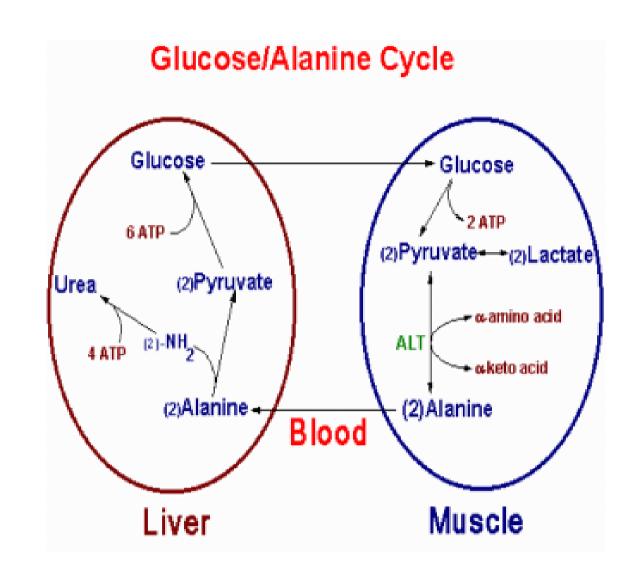
#### **Muscle-derived lactate:**

- During anaerobic muscle activity (hypoxia), lactate is produced in the muscles, released into the bloodstream, and transported to the liver, where it is converted into glucose.
- This **glucose** is then made available again to the **muscles**.
- This cycle of glucose-lactate is known as the Cori cycle.



# Gluconeogenesis from Alanine of Muscular Origin

- A constant precursor of glucose during prolonged fasting, alanine originates from the catabolism of muscle proteins as well as from glycolytic pyruvate.
- Pyruvate is converted into alanine.
- Alanine is then released into the bloodstream, transported to the liver, and subsequently reconverted into pyruvate.
- This **glucose-alanine** cycle is known as the **Felig cycle**.

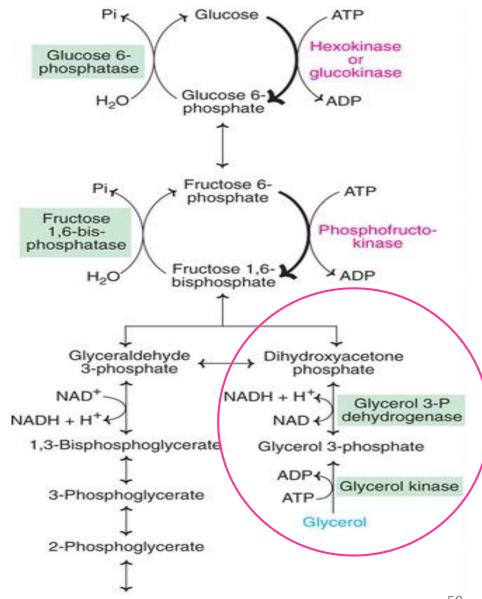


## Gluconeogenesis from glycerol

- ➤ Glycerol is the degradation product of **triglycerides** (from dietary sources, circulating lipoproteins, and adipose tissues)
- ➤ The **liver** and kidneys possess *glycerol kinase*, which phosphorylates glycerol into **glycerol-3**-**phosphate**.

#### This molecule can then:

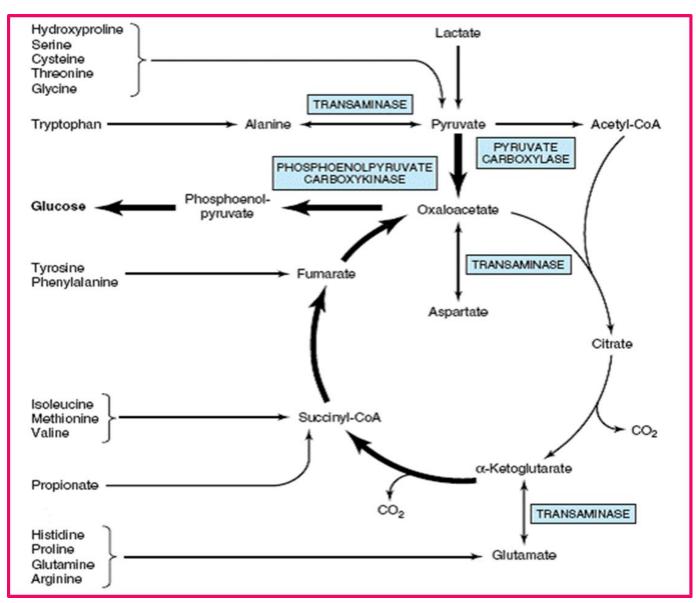
- Accept fatty acids (to synthesize triglycerides)
- ➤ Be **oxidized** into **dihydroxyacetone phosphate** by **glycerol-3-phosphate dehydrogenase** and enter the **gluconeogenesis** pathway.



## Gluconeogenesis from glucogenic amino acids

Amino-acids whose carbon skeleton is converted into pyruvate or into one of the four intermediates of the Krebs cycle (α-ketoglutarate, succinyl-CoA, fumarate, or oxaloacetate) are called glucogenic.

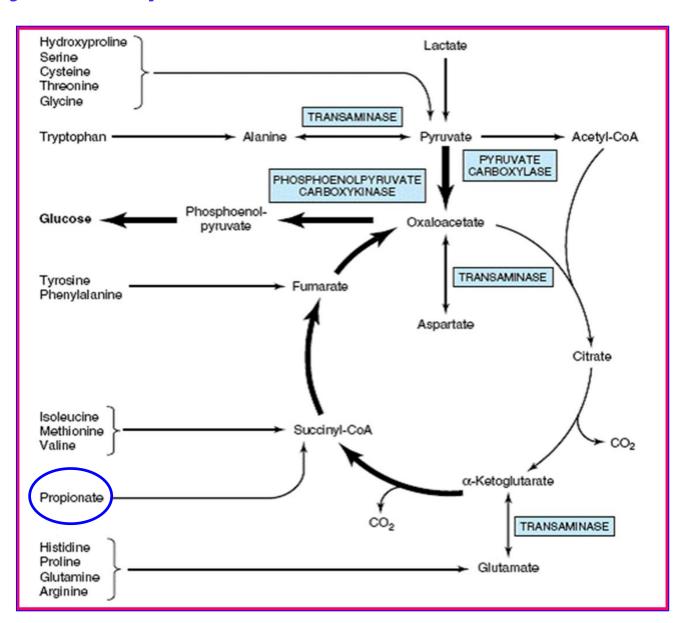
This carbon skeleton exits at the malate level and proceeds toward phosphoenolpyruvate (PEP).



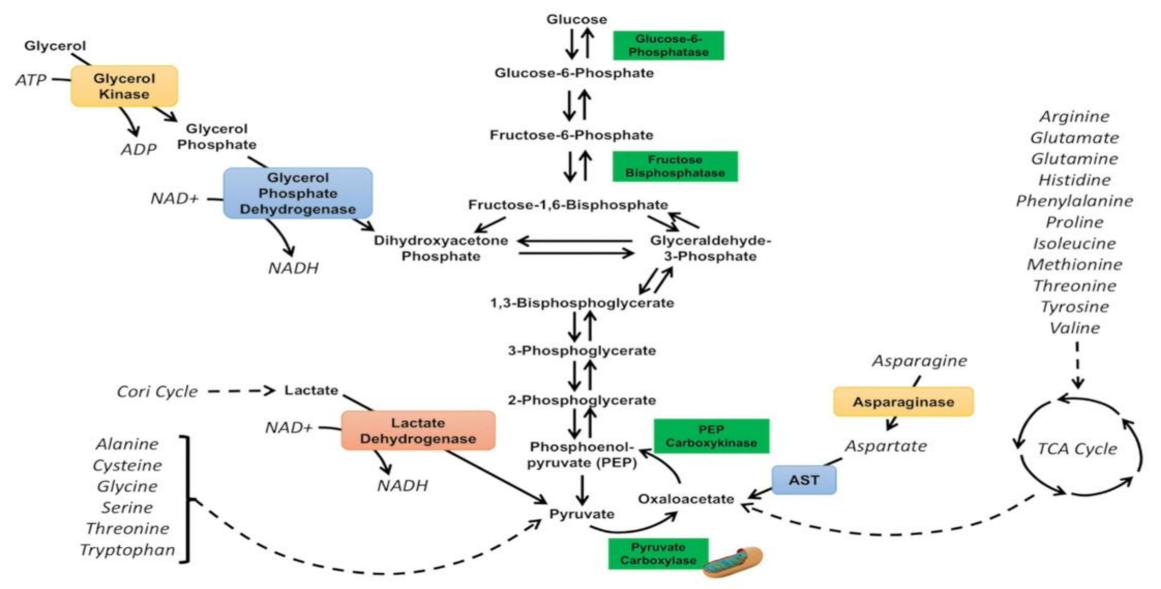
## Gluconeogenesis from Propionate

ightharpoonup Propionate (CH<sub>3</sub>-CH<sub>2</sub>-COOH) is the degradation product of odd-chain fatty acids.

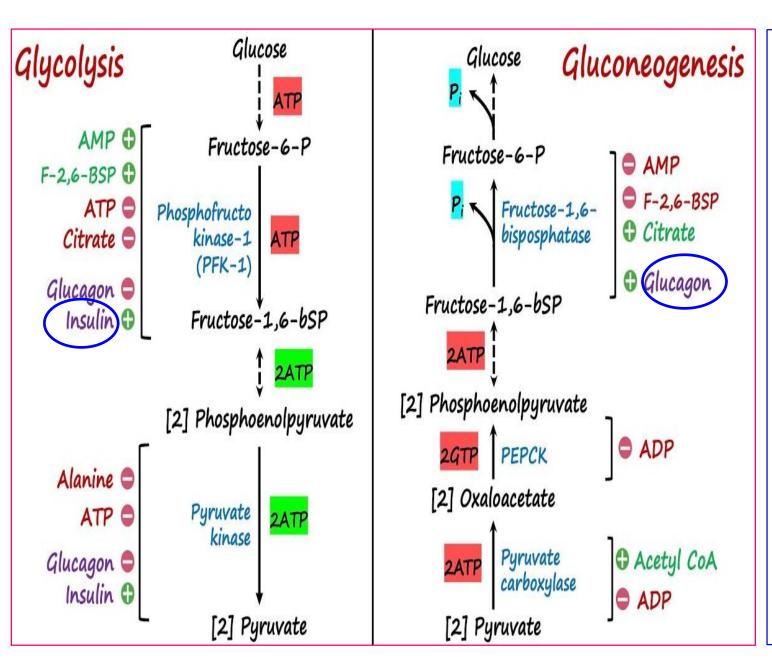
➤ It is **converted** into succinyl-CoA, an intermediate of the **Krebs cycle**, from which it can be directed toward the formation of **phosphoenolpyruvate** (PEP).



# Substrates and Enzymatic Steps of Gluconeogenesis



# Regulation of Gluconeogenesis



#### **Hormonal Regulation**

- ➤ During fasting periods (between meals), blood glucose levels decrease, leading to the secretion of glucagon by the endocrine pancreas.
- ➤ Glucagon accelerates gluconeogenesis and inhibits glycolysis. Additionally, glucagon induces the synthesis of key gluconeogenic enzymes, including:

Phosphoenolpyruvate carboxykinase (PEPCK) Fructose-1,6-bisphosphatase (F1,6BPase).

➤ Conversely, insulin has the opposite effect: during the postprandial period (after a meal), it slows down gluconeogenesis and activates glycolysis.

# Glycogen Metabolism

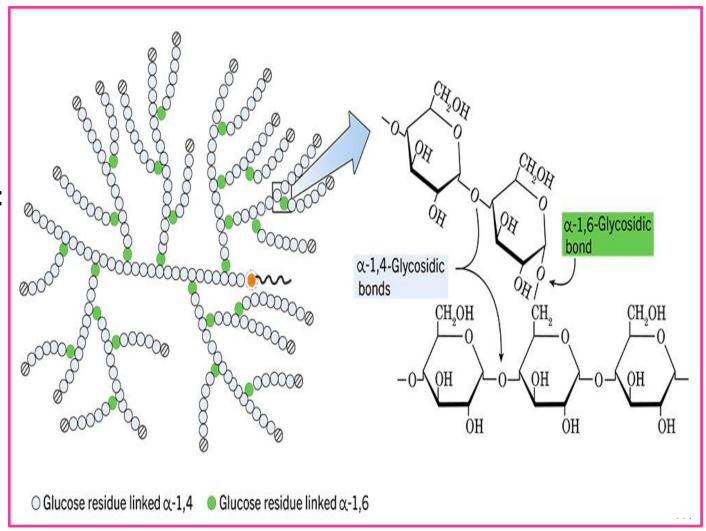
#### Introduction

Glycogen metabolism, whether synthesis

(glycogenesis) or breakdown

(glycogenolysis), is regulated according to:

- The body's nutritional state (fed or fasting)
- > Cellular energy status (ATP/AMP ratio).



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#### **Introduction**

Glycogen Metabolism occurs in the intestines, liver, and muscles.

- ➤ In the intestines: During the postprandial period, the digestive catabolism of glycogen and, more importantly, dietary starch produces glucose, which is directed to:

  Storage sites, in the form of glycogen (liver and muscles)

  Sites of consumption, where it is used as an energy substrate
- ➤ In the liver: During the postprandial period, intestinal glucose and glucose produced between meals via gluconeogenesis are stored as glycogen through glycogenesis.

  During fasting periods, glucose derived from glycogenolysis is exported to consuming tissues.

  The hepatic glycogen store, which can be depleted within 24 hours, is considered "public use".

#### > In the muscles:

At rest, glucose is stored as glycogen through glycogenesis.

During muscular activity, glycogenolysis provides an immediate source of glucose, which is used locally as an energy substrate. Muscle glycogen is considered "private use".

## Pathways of Glycogenesis and Glycogenolysis

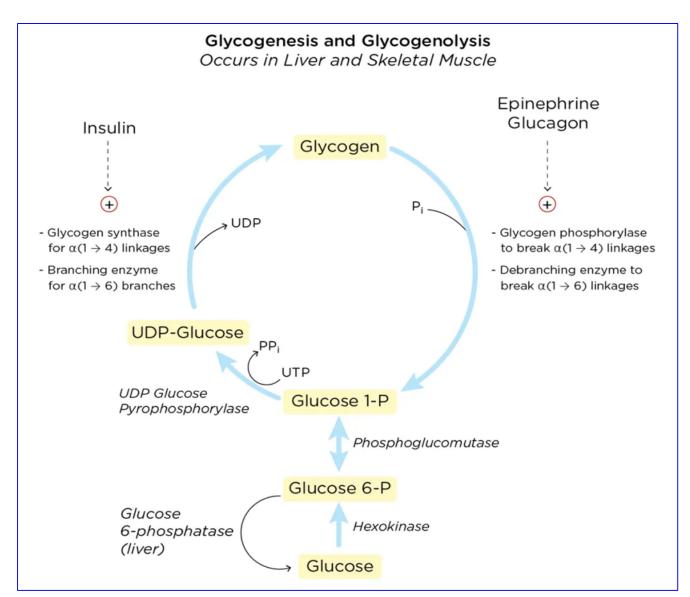
#### Glycogenesis (Glycogen Synthesis ):

The process of **forming glycogen** from **glucose**. Catalyzed primarily by *glycogen synthase* and *branching enzyme*. **Stimulated** by **insulin** during the **postprandial period**.

#### Glycogenolysis (Glycogen Degradation ):

The **breakdown** of **glycogen** into **glucose-1-phosphate**, which can be converted to **glucose-6-phosphate**. Catalyzed by *glycogen phosphorylase* and *debranching enzyme*. **Stimulated** by **glucagon** (in **liver**) and **epinephrine** (in muscle) during **fasting** or **exercise**.

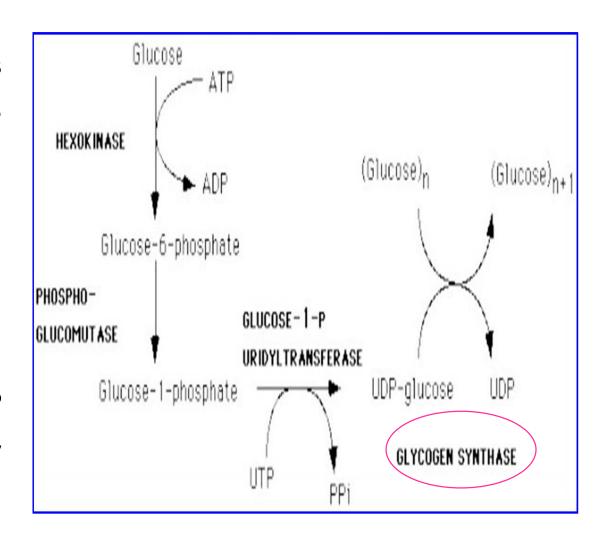
These pathways are tightly regulated to maintain blood glucose homeostasis and provide energy according to the body's nutritional and energy status.



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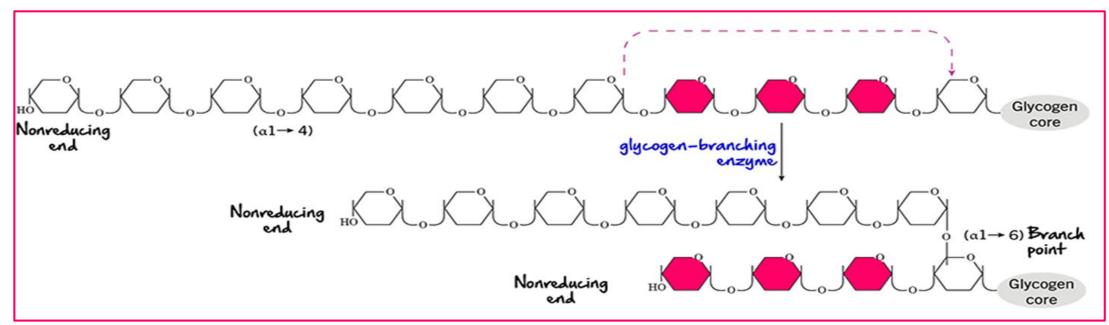
# Glycogenesis: Glycogen Synthesis

- The purpose of **glycogen synthesis** is to store **excess** glucose in the liver (approximately 150 g, one-third of the body's total glycogen) and muscles (approximately 300 g, two-thirds of the total glycogen).
- Glycogen synthesis occurs in the cytosol.
- The key enzyme is *glycogen synthase*.
- The precursor is **glucose-6-phosphate** (G6P).
- The synthesis process requires **energy** in the form of **ATP** and **UTP** (uridine triphosphate), consuming **two** high-energy **bonds**.

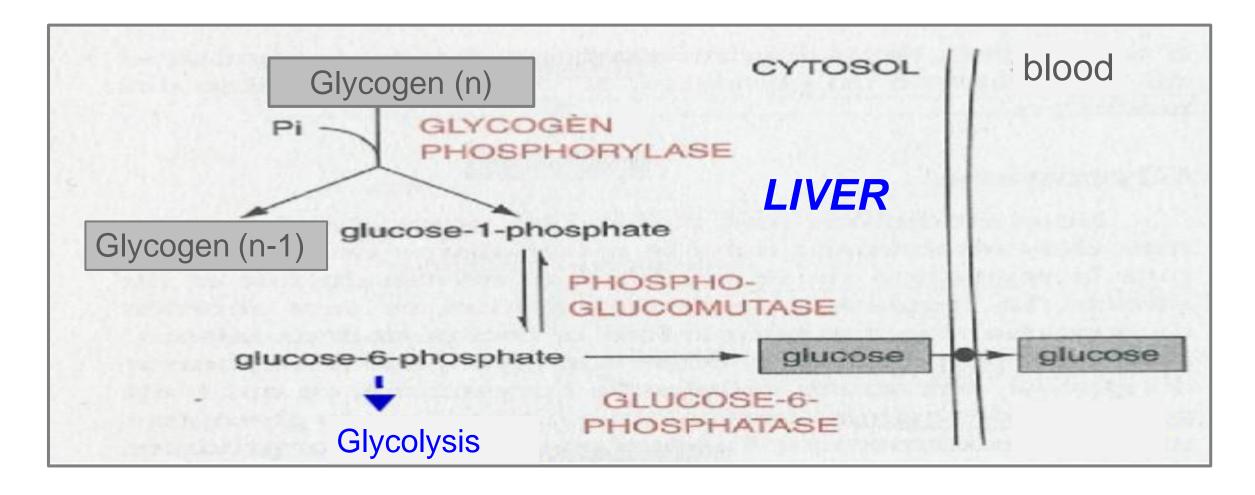


## **Branch Formation in Glycogen**

- When a **glycogen** chain elongates by at least **11 glucose residues**, the branching enzyme cleaves an  $\alpha(1\rightarrow 4)$  bond and **releases** a fragment of **7 glucose residues**. This fragment is then transferred to the **C6 position** of a glucose residue on the same or another chain, forming an  $\alpha(1\rightarrow 6)$  linkage.
- This process results in a highly branched glycogen structure, which enhances its **solubility** and provides multiple sites for rapid addition or removal of glucose units.

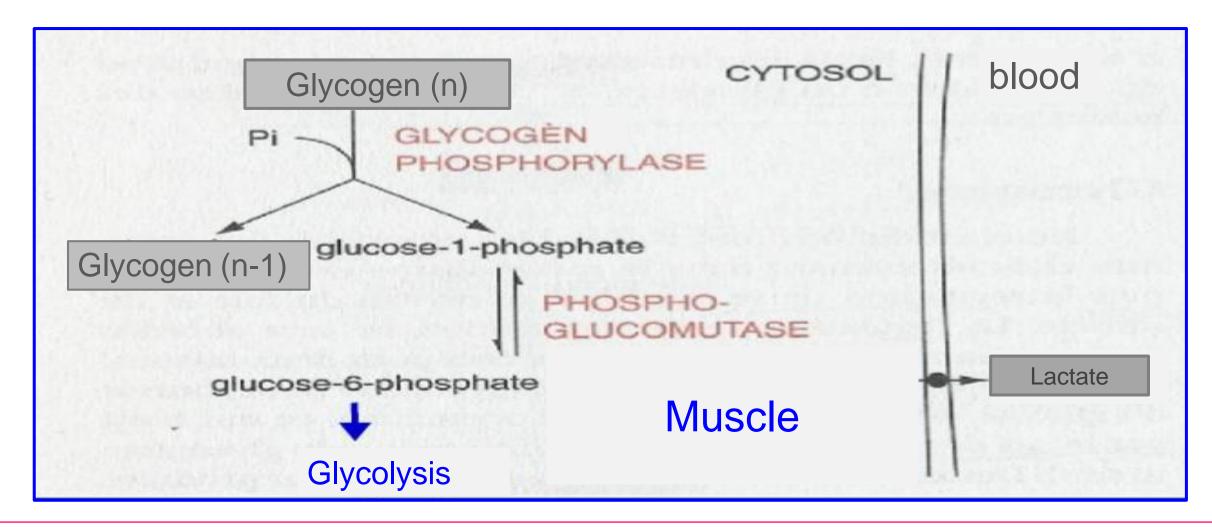


## Glycogenolysis: Glycogen Degradation (Liver)



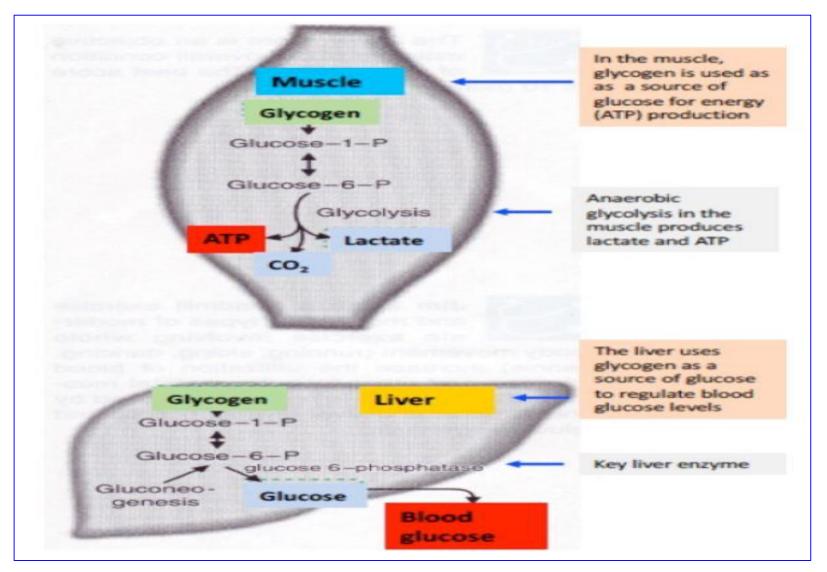
**Note:** Glucose-6-phosphatase is a liver-specific enzyme that hydrolyzes glucose-6-phosphate to free glucose, allowing the liver to export glucose into the bloodstream and maintain blood glucose homeostasis.

# Glycogenolysis: Glycogen Degradation (Muscle)



**Note:** The **glycogen stored** in muscle is restricted to **local use**, as muscle cells do not express **glucose-6-phosphatase**. Therefore, glucose-6-phosphate generated from glycogen cannot be converted to free glucose and exported; it is **used intracellularly** for **energy production.**DAROUI-MOKADDEM H.

## Use of Hepatic and Muscular Glycogen



# **Glycogen storage diseases**

Туре	Name	Enzyme	Glycogen	Prevalence	Clinical
0		Glycogen synthase	None	Very rare	Hypoglycemia, ketosis
1	VonGierke	Glucose-6-phosphatase	Normal	1/50000	Severe hypoglycemia, big liver, lactic acidosis
11	Pompe	Lysosomal a1,4-glucosidase	Inclusion bodies with glycogen	1/140000	Big heart, weakness, death
Ш	Cori	Debranching enzyme (a1,6- glucosidase)	Shorter outer branches	1/100000	Mild hypoglycemia, big liver
IV	Anderson	Branching enzyme	Long chains with few branches	Very rare	Cirrhosis, hypoglycemia, death
V	McArdle	Muscle glycogen phosphorylase	Normal	1/100000	Muscle cramps, weakness, teens
VI	Hers	Liver glycogen phosphorylase	normal	1/70000	Mild hypoglycemia, cirrhosis